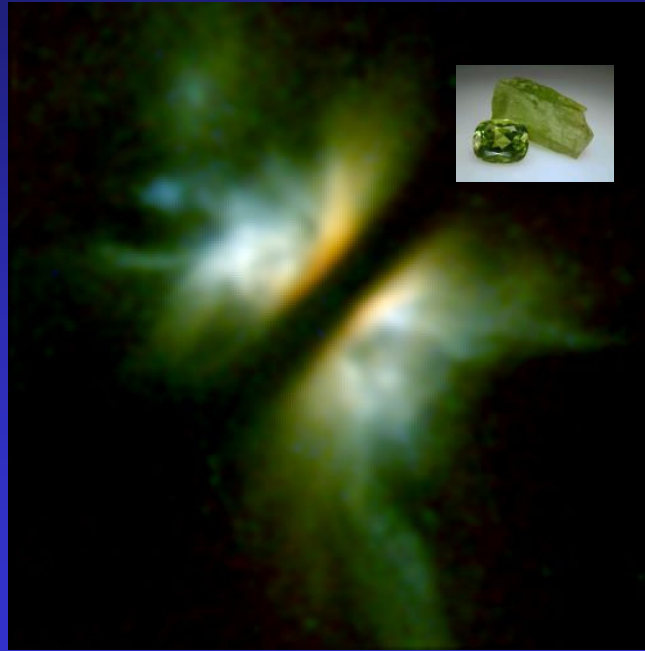


# Mineralogy of Protoplanetary Dust

Thomas Henning  
Max Planck Institute for Astronomy  
Heidelberg



Disks 2006, September 2006, Vidago Palace, Portugal

## Motivation

### Mineralogy of protoplanetary dust



- Infrared spectroscopic properties as a diagnostic tool (Optical depth, temperature, chemistry, growth processes, mixing, ...)
- Grains: Surface for chemical reactions and opacity source (Important refractory condensates)
- Interesting structural and optical behaviour (Tunneling processes at low temperatures)

# Metal Oxides – Important Class of Materials



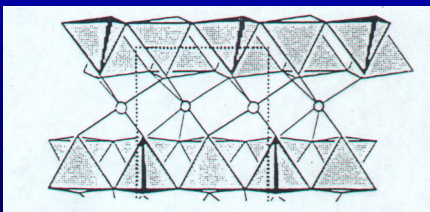
Abundances:  $A_N(X) = 12 + \text{Log} (N_X/N_H)_{\text{sun}}$

- Oxygen:  $\sim 8.9$
- Mg, Si, Fe:  $\sim 7.50$
- Al, Ca:  $\sim 6.50$

## Structure of Silicates

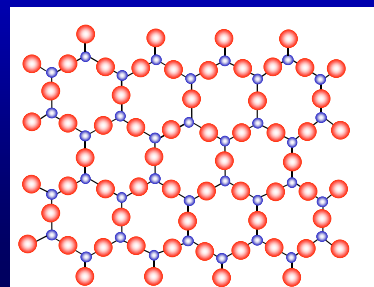
**Pyroxenes: solid solution series**  
**MgSiO<sub>3</sub>-FeSiO<sub>3</sub>**

Enstatite - Ferrosilite



chain  
silicates  
 $[\text{SiO}_3]_{\infty}^{2-}$

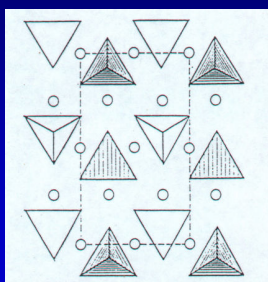
**Crystalline silica**



layer  
silicates  
SiO<sub>2</sub>

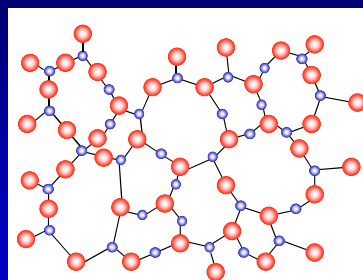
**Olivines: solid solution series**  
**Mg<sub>2</sub>SiO<sub>4</sub>-Fe<sub>2</sub>SiO<sub>4</sub>**

Forsterite -Fayalite



neso silicates  
 $[\text{SiO}_4]^{4-}$

**Amorphous silica**

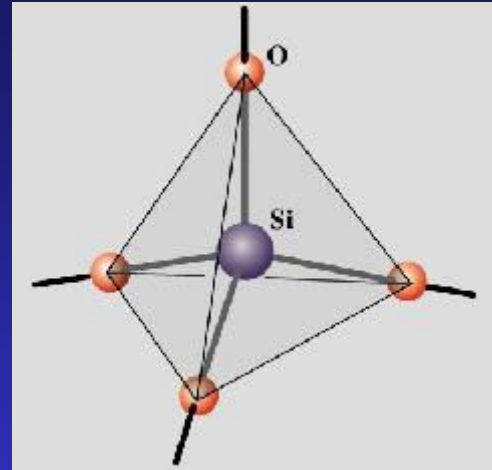
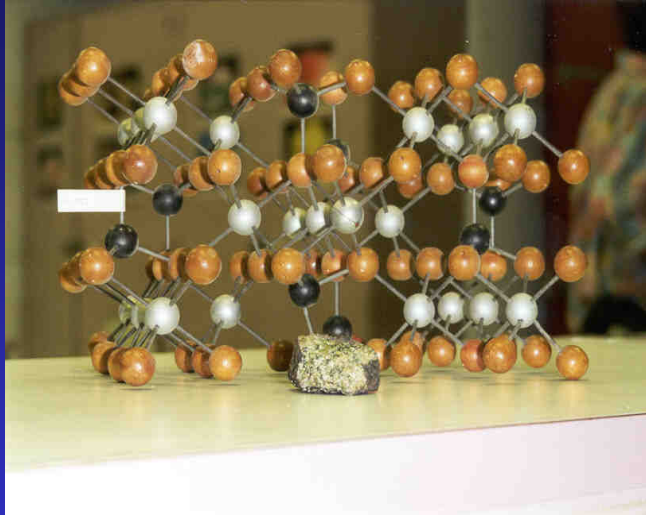


3-dim.  
network

Incorporation  
of modifier  
cations destroys  
oxygen bridges

 non-bridging oxygen

# Silicate Tetrahedron – A Basic Unit



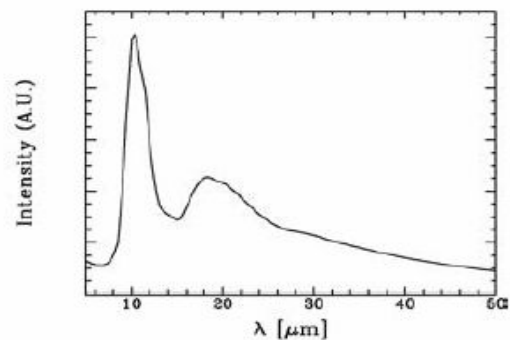
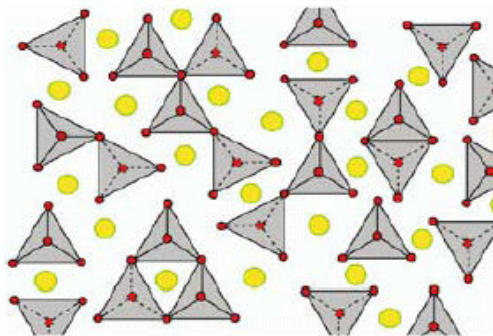
•  $\text{SiO}_4$  tetrahedron: Radius of central  $\text{Si}^{4+}$  ion is 0.041 nm

Radius of the  $\text{O}^{2-}$  ion is 0.140 nm

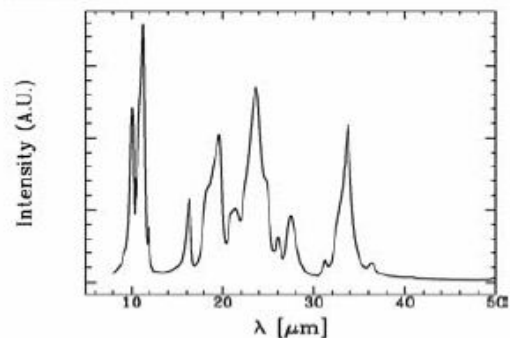
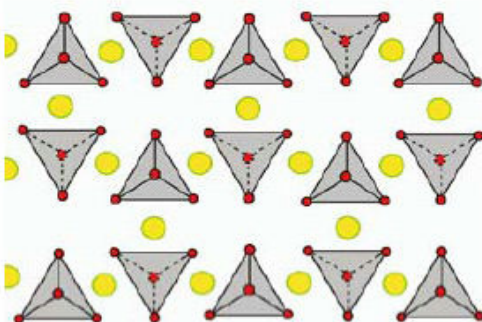
The ratio of these radii is  $= 0.041/0.140 = 0.3 \rightarrow$  Volume is basically filled with oxygen atoms; Number of its nearest equal neighbours (coordination number) is 4  $\rightarrow$  Nearest neighbours form a tetrahedron

## IR Properties of Silicates – Amorphous vs. Crystalline Structures

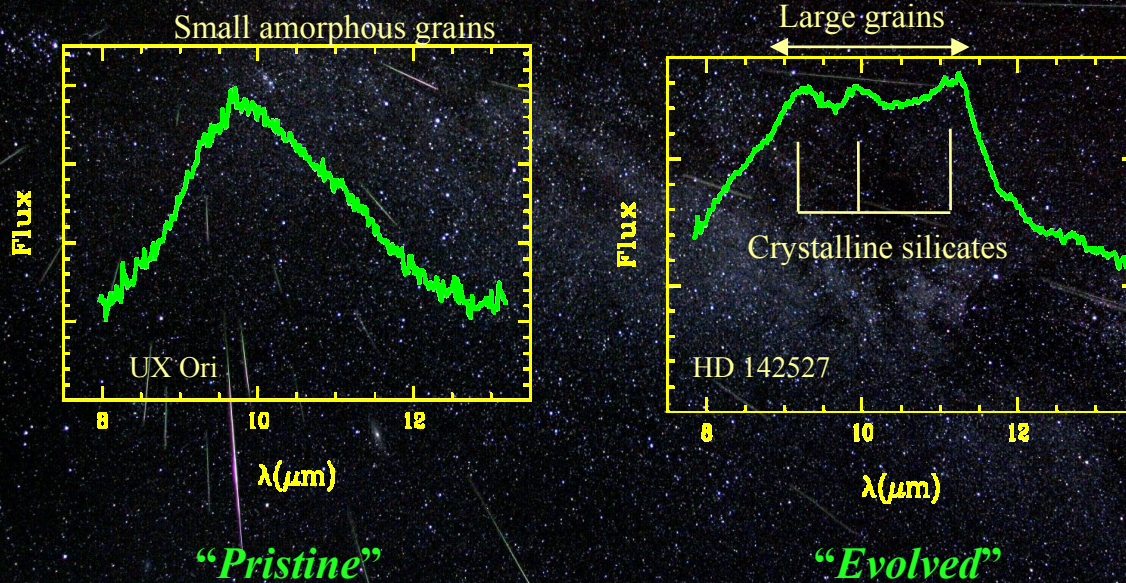
**Amorphous structure**



**Crystalline structure**



# Dust processing at 10 micron



## IR Properties of Silicates – Amorphous vs. Crystalline Structures

- 10 $\mu\text{m}$  band due to Si-O stretching; position depends on level of  $\text{SiO}_4$  polymerization (e.g. band shifts from 9.0  $\mu\text{m}$  for  $\text{SiO}_2$  to 10.5  $\mu\text{m}$  for  $\text{Mg}_{2.4}\text{SiO}_{4.4}$  – Jäger et al. 2003)
- 18  $\mu\text{m}$  band additionally broadened (coupling of the Si-O bending to the Me-O stretching vibration)
- Crystalline silicates: Bands beyond 20  $\mu\text{m}$  caused by translational motion of metal cations within the oxygen cage and complex translations involving Me and Si atoms

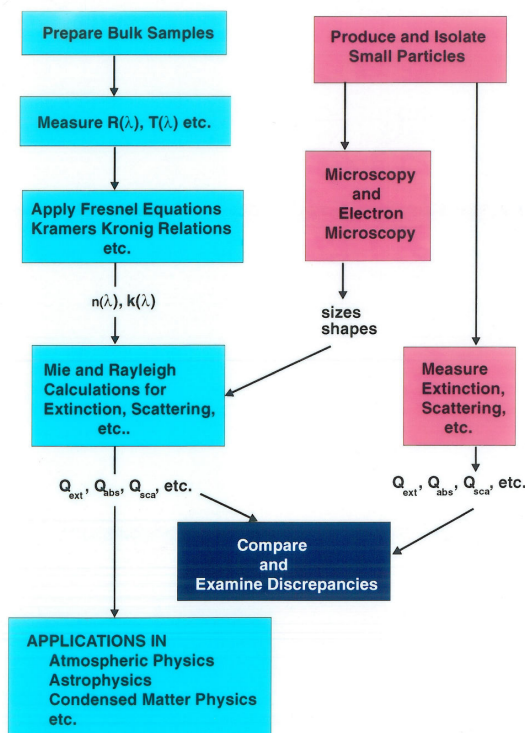
# Basic Optical Data Cosmic Dust Analogues

- Broad Wavelength Range
- Appropriate Structure  
(Fe/Mg, am./cryst. ...)
- Isolated Small Particles
- Temperature Range

Jena database of optical data (Henning et al. 1999)

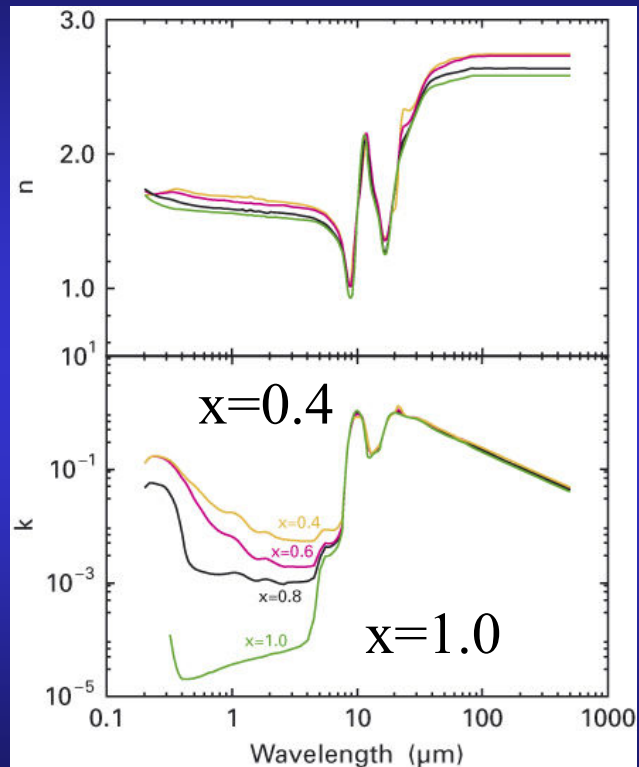
<http://www.astro.uni-jena.de/Group/Subgroups/Labor/odata.html>

## Laboratory measurements



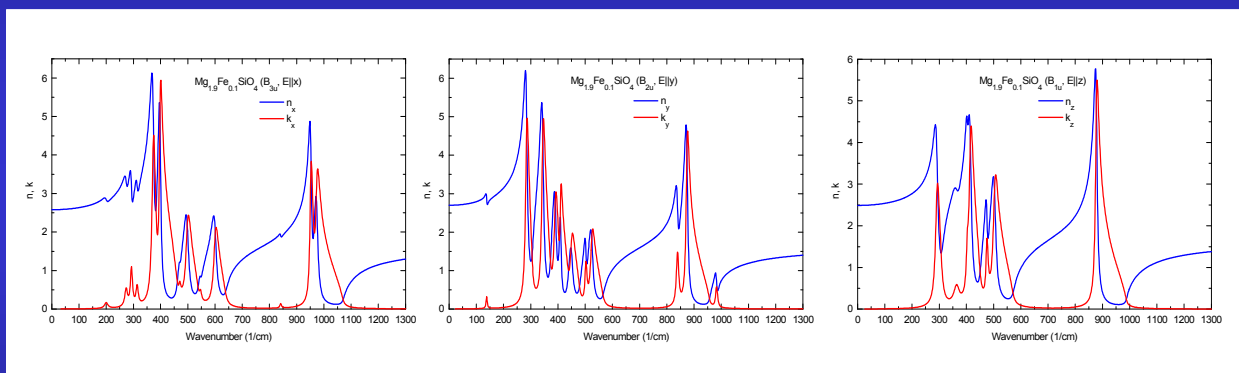
# Optical data of Amorphous Pyroxenes $Mg_xFe_{1-x}SiO_3$

Increase  
of NIR absorptivity  
with Fe content



(J. Dorschner, B. Begemann, Th. Henning, C. Jäger and H. Mutschke, A&A 1995)

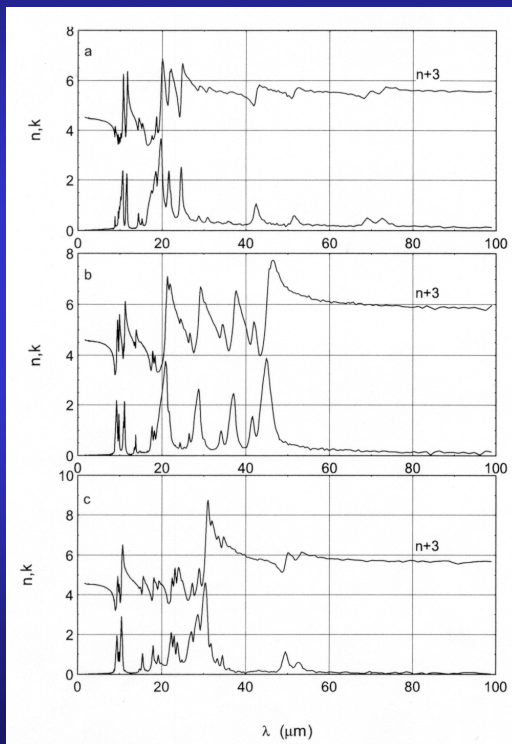
## Complex refractive index $n + ik$ : Olivine Crystal



Fabian, Henning et al. (2001)

# Basic Optical Data for Silicates

Enstatite



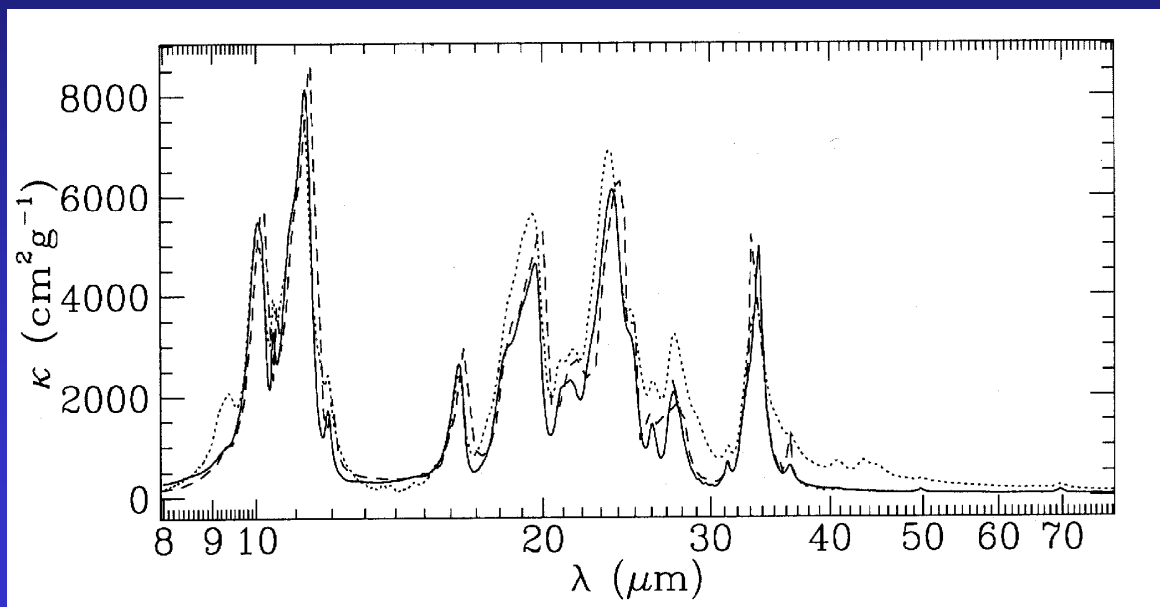
Jäger et al. (1998)

*Steps toward interstellar  
silicate mineralogy*  
Jena 1994-2003

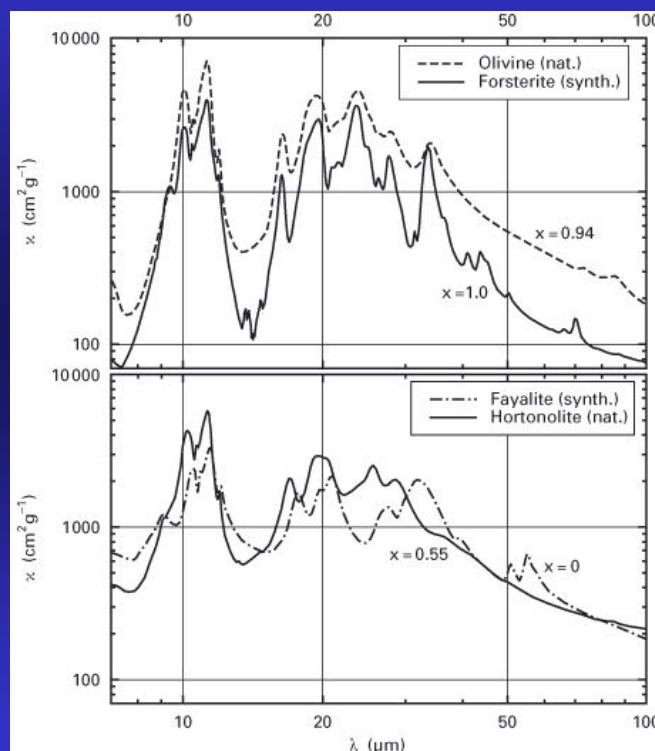
See also:  
Papers by Neaples and Kobe  
Groups

## Forsterite $Mg_2SiO_4$

- Two strong features in the 10 micron region
- Two prominent features in the 18 micron region
- A strong band in the 23 micron complex
- Strong band in the 33 micron complex



## Mg<sub>2x</sub>Fe<sub>2-2x</sub>SiO<sub>4</sub> (Olivines)



Jäger et al. (1998)

## Increase of iron content

Olivines Mg<sub>2x</sub>Fe<sub>2-2x</sub>SiO<sub>4</sub>

- (1) Strengths of 10 and 20  $\mu\text{m}$  bands relative to the underlying continuum decrease
- (2) Band peak positions are shifted to longer wavelengths ( $\Delta\lambda\alpha\lambda^2$ ; bands are shifted by the same amount in wavenumber) – growing metal-oxygen distances
- (3) 33.6 and 33.8  $\mu\text{m}$  features: only shoulders in hortolonite ( $x = 0.55$ ), disappear for fayalite ( $x = 0.0$ )

Pyroxenes Mg<sub>x</sub>Fe<sub>1-x</sub>SiO<sub>3</sub>

- (1) Similar behaviour as olivines with the exception of the 10  $\mu\text{m}$  region (Si-O stretching modes not sensitive to Fe incorporation)



# LOW-TEMPERATURE EFFECTS

Henning and Mutschke (1997)

## Crystalline Dielectric Solids

- IR bands (single phonon transitions):  
Sharpening because of decreased damping, shift to shorter wavelengths
- FIR absorption (phonon difference processes):  
significant reduction because of decreasing phonon number

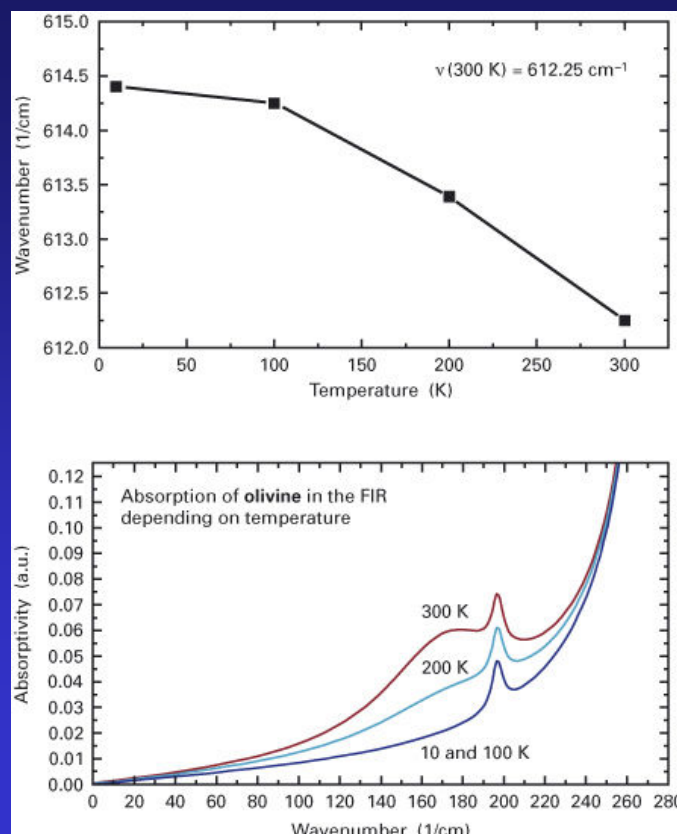
## Amorphous Dielectric Solids

- FIR absorption:  
Dominated by disorder-induced single phonon processes, no temperature dependence
- Millimeter range:  
highly temperature-dependent low energy processes, e.g. tunneling transitions in glasses

## Semiconductors

- free charge carrier absorption:  
vanishes because conduction band is depopulated

## Temperature Behaviour of Optical Properties (Olivine)



What is expected ?

=> Bands are broadened and shifted to lower frequencies with higher temperature

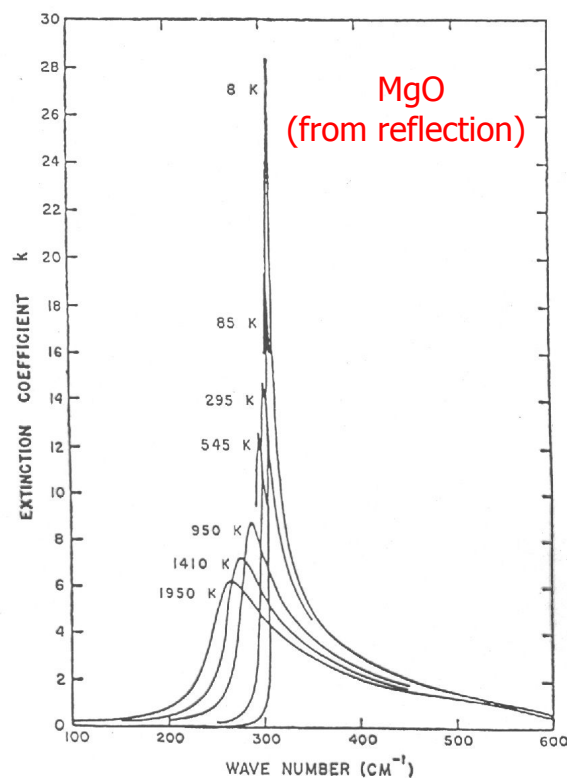
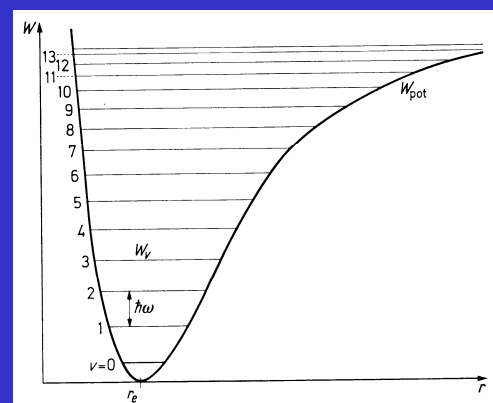
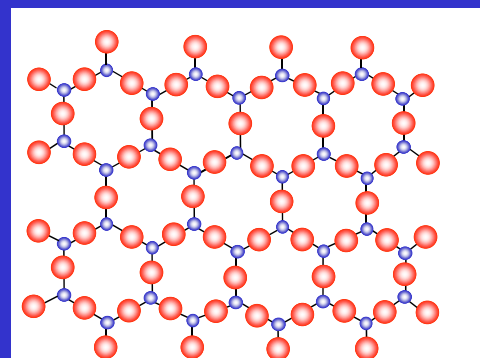


Fig. 9. Computed extinction coefficients of MgO as functions of wave number and temperature. (From Jasperse *et al.* [12].) 1966

Henning & Mutschke 1997, Mennella *et al.* 1998, Bowey *et al.* 2001, Chihara *et al.* 2001, Koike, Mutschke *et al.* 2005

### Explanation (higher T):

- Lattice expansion => smaller forces => lower excitation frequencies
- Moving atoms => broader frequency range for excitation
- physically: anharmonicity of potential + scattering at other phonons (shorter lifetime – broader levels)

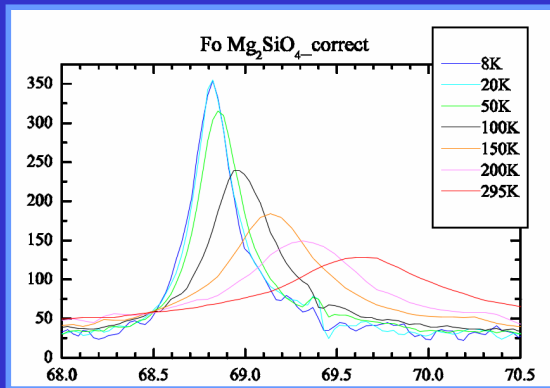


# Band changes at low temperatures

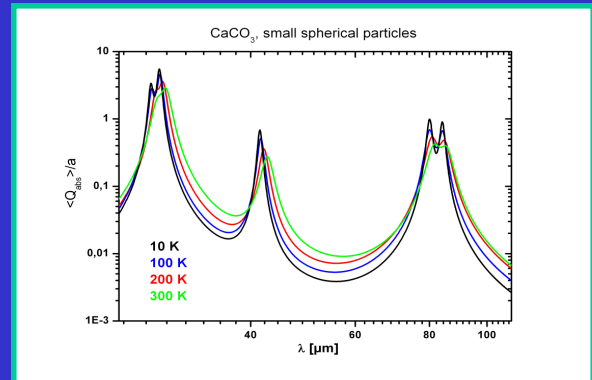
Prominent examples:

69 $\mu$ m forsterite ( $Mg_2SiO_4$ ) band

41-43 $\mu$ m [spher. grains] calcite ( $CaCO_3$ ) band

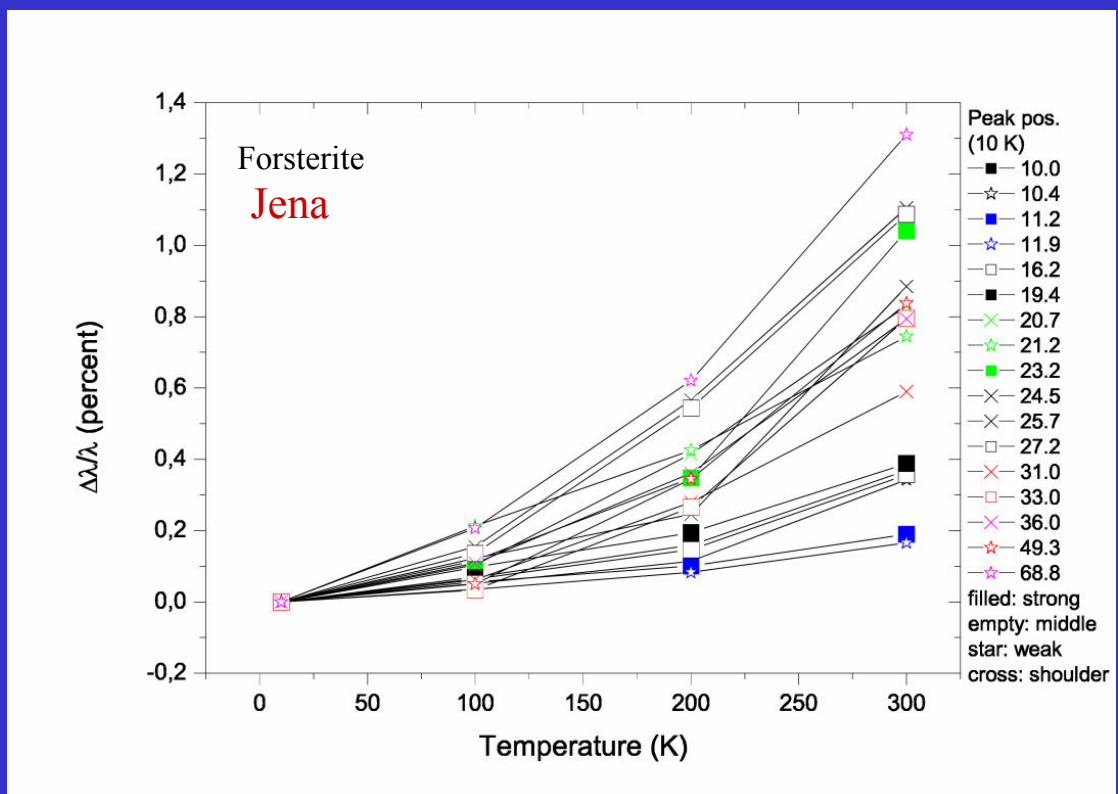


Koike, Mutschke et al. (2006)

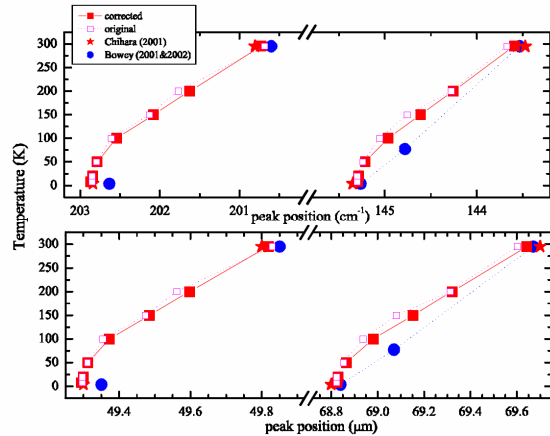
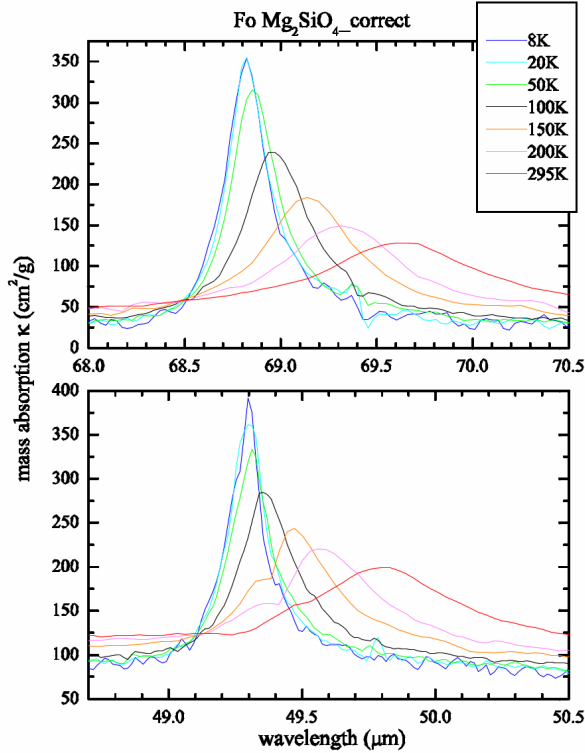


Posch, Mutschke et al. (2006)

## How big is the relative peak shift ?

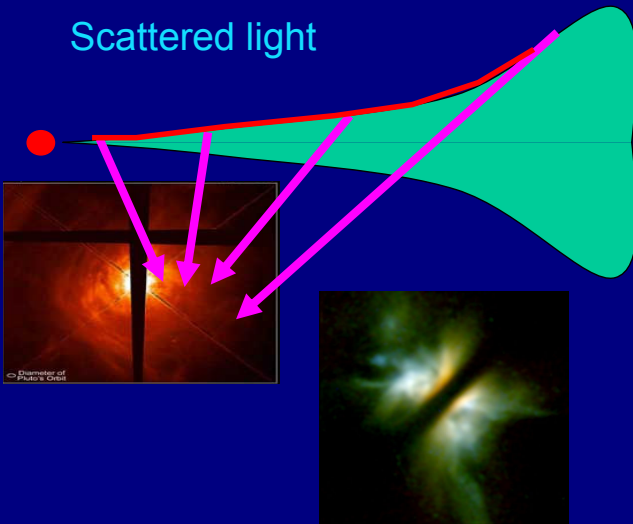


# Long-wavelength forsterite bands as thermometer

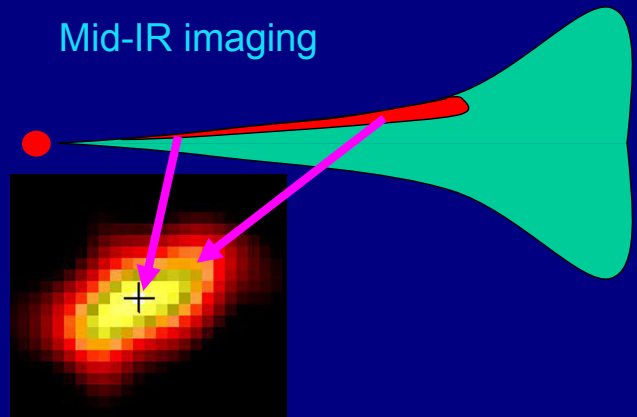


# Which observations probe which grains?

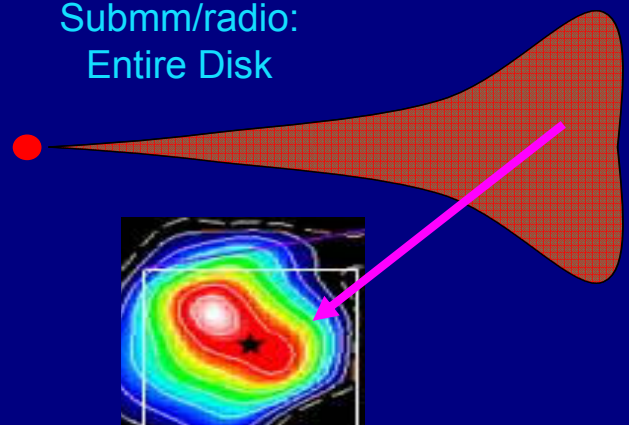
Scattered light



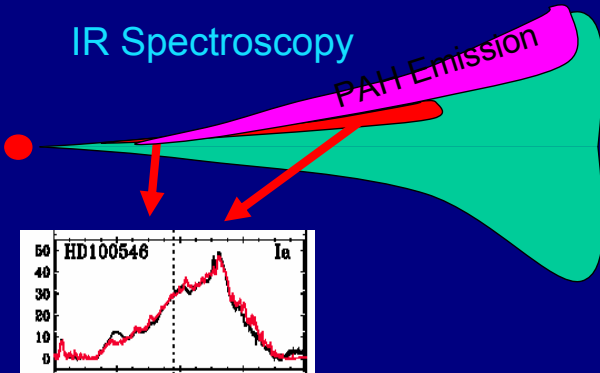
Mid-IR imaging



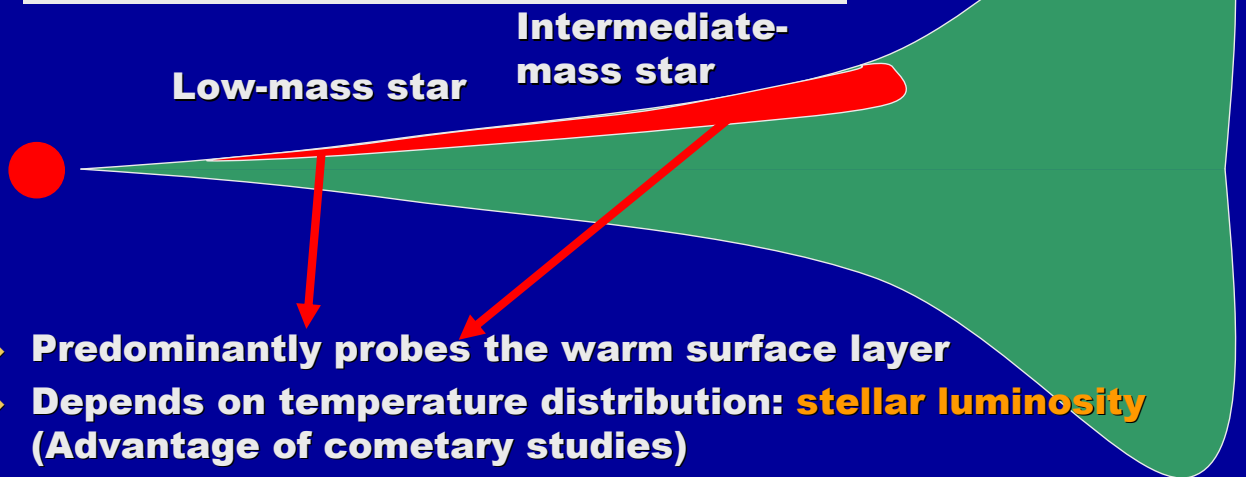
Submm/radio:  
Entire Disk



IR Spectroscopy

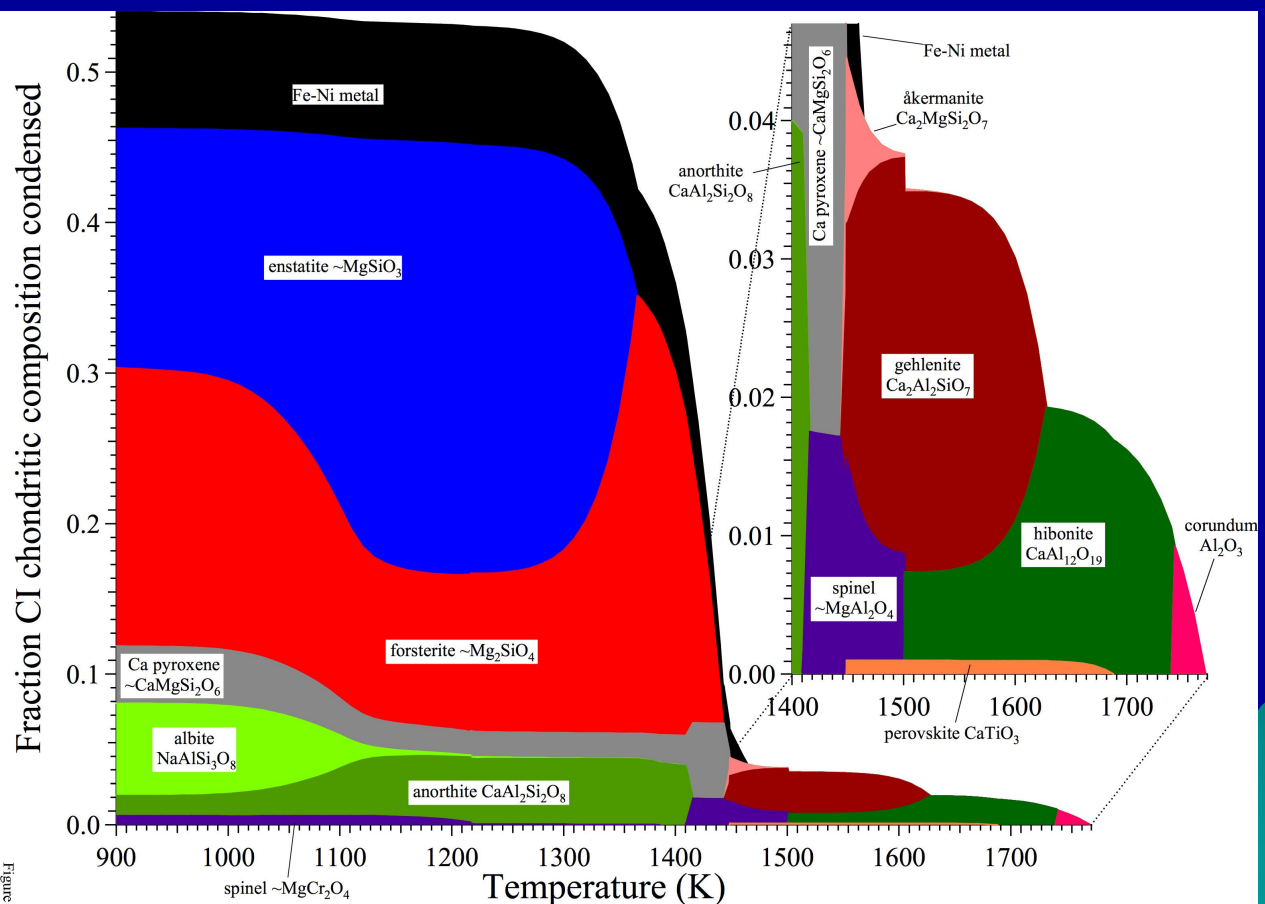


# IR spectroscopy: Limitations



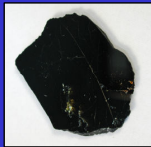
- ◆ Predominantly probes the warm surface layer
- ◆ Depends on temperature distribution: **stellar luminosity** (Advantage of cometary studies)
- ◆ Trace grains only up to micron sizes
- ◆ Lower sensitivity compared to millimeter spectroscopy
- ◆ Ground-based: Small wavelength range
- ◆ Space-based: Low spatial resolution

## Dust condensation sequence



# Metal Oxides in the Dust Condensation Sequence

For solar elemental abundances and a pressure of  $5 \cdot 10^{-9}$  bar, the following *stability limits* of oxides and silicates containing abundant elements are currently predicted:



Corundum  $\alpha\text{-Al}_2\text{O}_3$ : ~ 1420K

Hibonite  $\text{CaAl}_{12}\text{O}_{19}$ : ~ 1320K

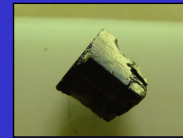
Perovskite  $\text{CaTiO}_3$ : ~ 1300K

Gehlenite  $\text{Ca}_2\text{Al}[(\text{Si},\text{Al})_2\text{O}_7]$ : ~ 1200K

Grossite  $\text{CaAl}_4\text{O}_7$ : ~ 1200K

Spinel  $\text{MgAl}_2\text{O}_4$ : ~ 1150K

Forsterite  $\text{Mg}_2\text{SiO}_4$ : ~ 1090K



(Gail 2003;  
Ebel&Grossman 2000)



## Dust Disk Mineralogy - Wavelength Positions



Forsterite 10.0, 11.3, 16.3, 23.5,  
27.5, 33.5, 69.7  $\mu\text{m}$

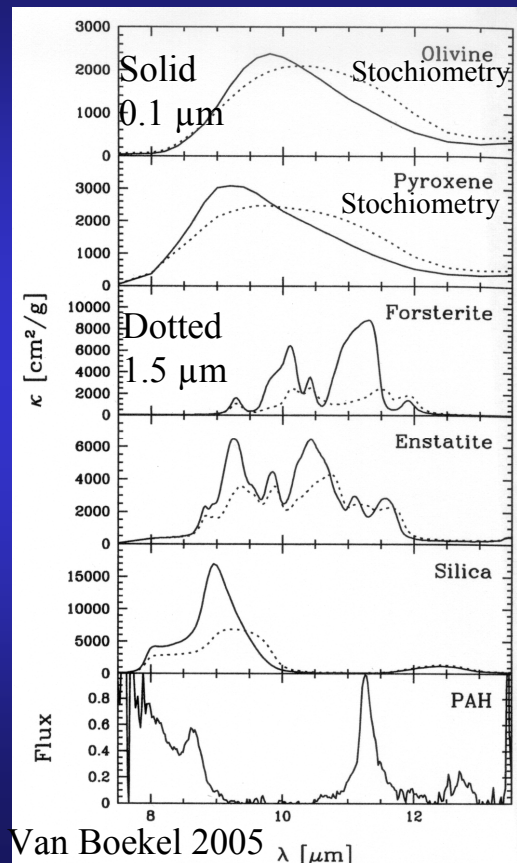


Enstatite 9.4, 9.9, 10.6, 11.1,  
11.6, 18.2, 19.3,  
21.5  $\mu\text{m}$

(strong features, exact positions  
vary with material + temp.)

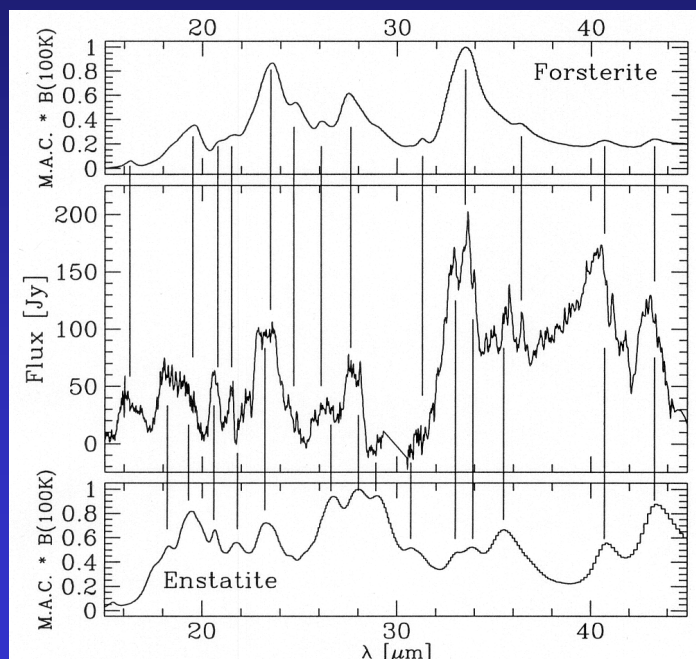
Koike et al. (1993, 2000, 2003),  
Colangeli et al. (1995),  
Jäger et al. (1994, 2003),  
Fabian et al. (2001),  
Chihara et al. (2002), ....

**Database: Henning et al. (1999)**



# Crystalline Revolution (ISO and Spitzer)

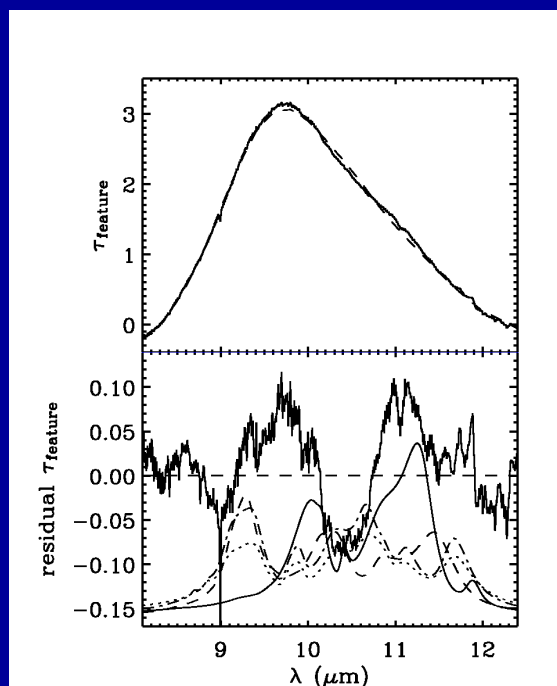
AFGL 4106



T=100 K

Jäger et al. (1998)

## Crystalline silicates in the ISM



Composition of amorphous silicates:

- Olivine ( $\text{MgFeSiO}_4$ ): 85%
- Pyroxene ( $\text{MgFeSi}_2\text{O}_6$ ): 15%

Crystallinity

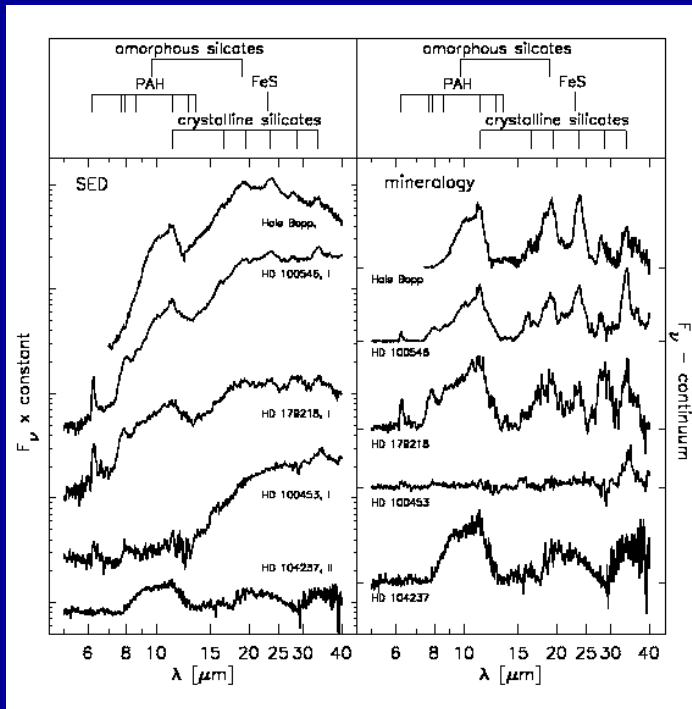
- **<0.4 %** of silicates in diffuse ISM are crystalline
- Crystallinity of **0.2% ( $\pm 0.2\%$ )** gives best fit to the 10 micron absorption feature

But:

Stellar ejecta are 10-20% crystalline!

(Kemper et al. 2004)

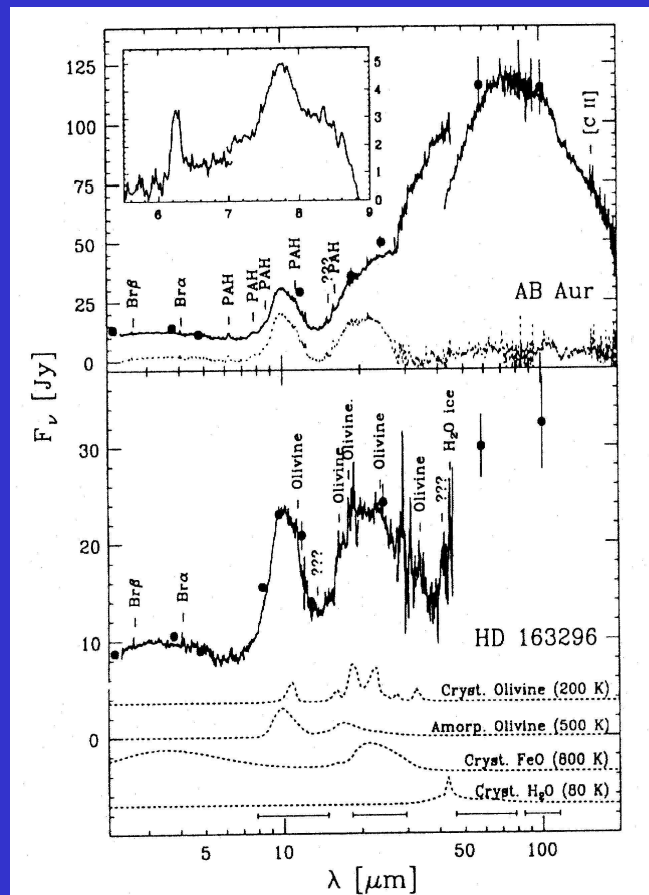
# ISO: Dust in Herbig Ae systems



- Dust mineralogy
  - Forsterite
  - Enstatite
  - FeS ???
  - Amorphous Silicates
  - PAH
  - Silica
- Slope change
- Grain growth

See Meeus et al. 2001, van den Ancker et al. 1999, Malfait et al. 1998, Bouwman et al. 2001, VandenBusche et al. 2002

## A few more examples

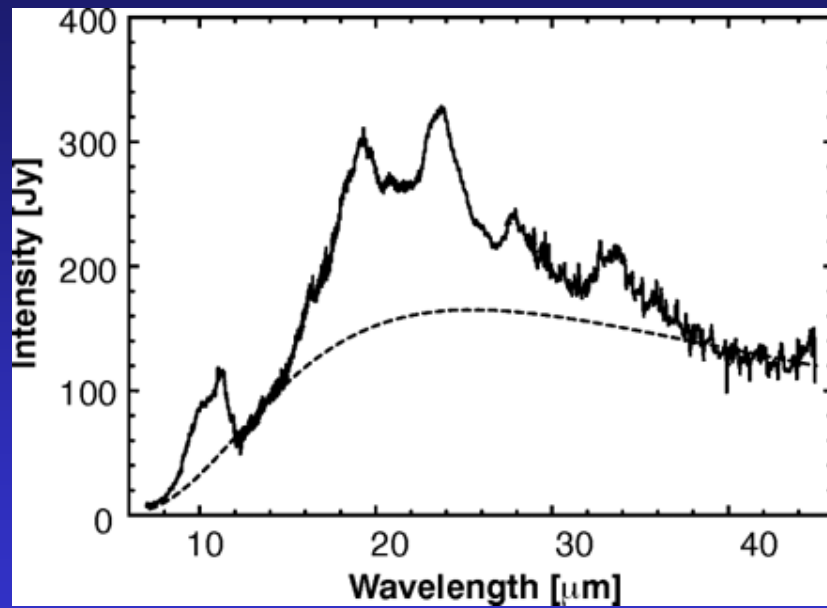


van den Ancker et al. (1999)



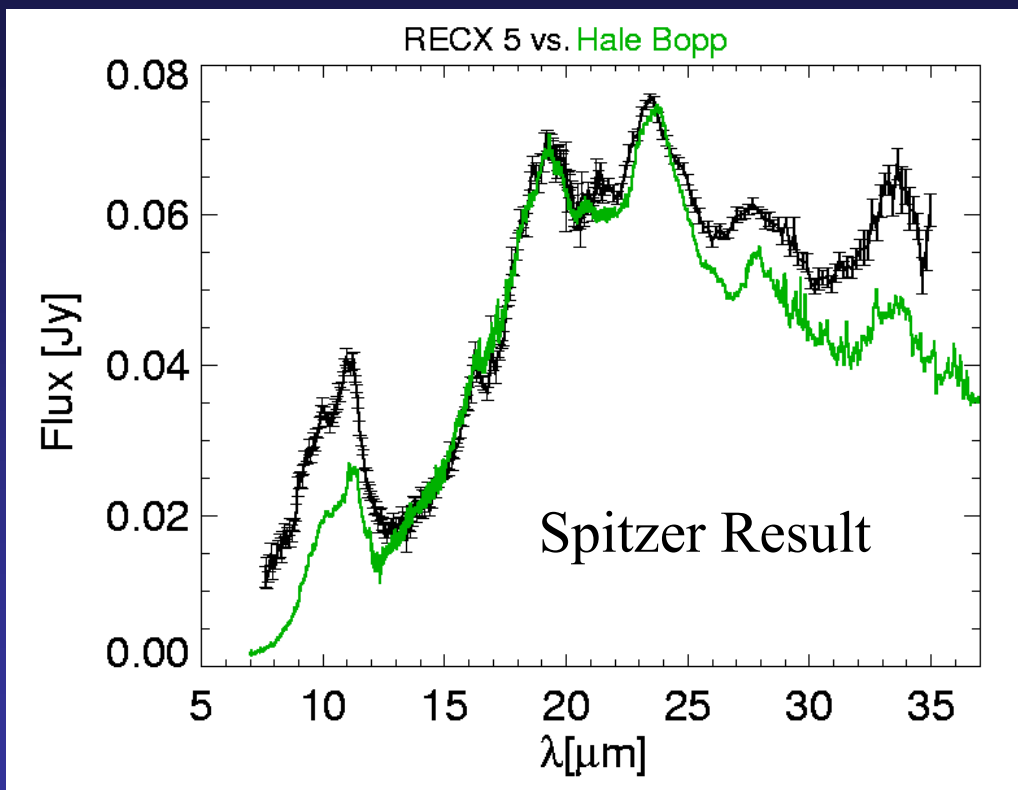
# Comet Hale-Bopp (C/1995 O1)

Crovisier et al.  
(1997), see also  
Wooden et al.  
(1999, 2000)



Evidence for crystalline Mg-rich olivines (forsterite)  
(e.g. 11.3, 16.5, 19.8, 24.0, 27.6, 33.9 micron bands)

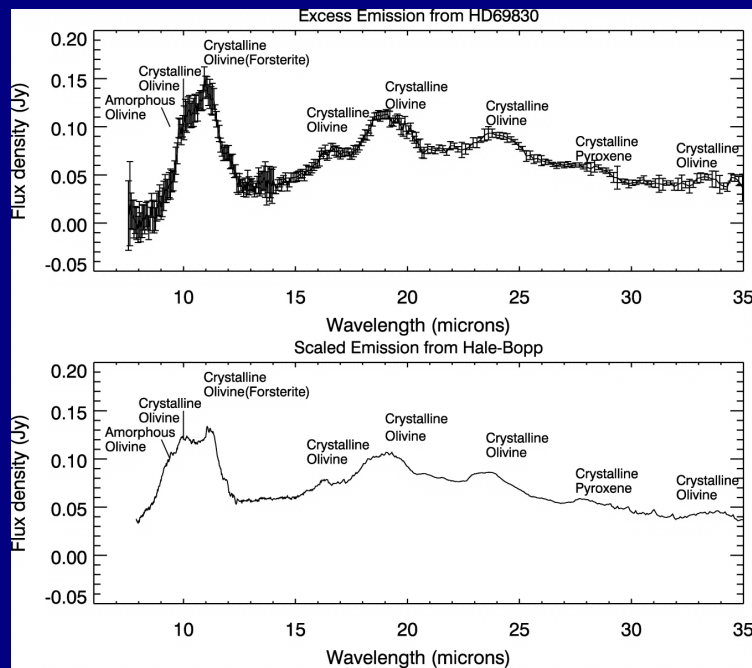
## RECX5: Hale Bopp Formation around an M4 star?



Bouwman, Lawson, Henning et al. (2006, in prep.)

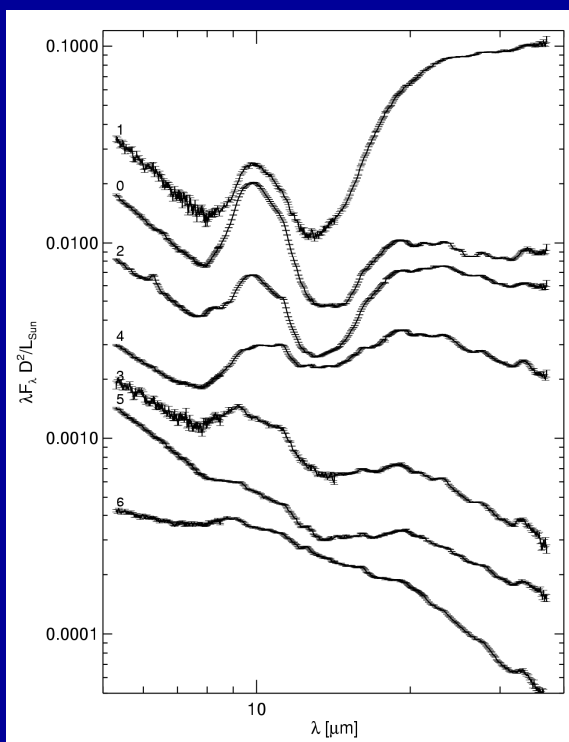
# Large Excess and Crystallinity in HD 69830 A Rare Case

(12.6 pc, K0V, 0.6-2 Gyr with higher age more probable)



Beichman  
et al. (2005)

## FEPS: 3-15 Myr old TTS stars

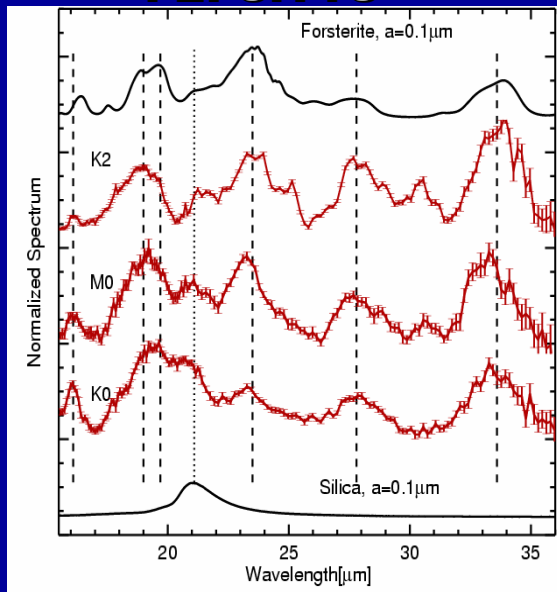


- Same as HAEBEs
- Strong variation in amorphous silicate band strength
- Changing SED slopes
- Unambiguous detection of PAHs

(Bouwman, Henning et al. 2006)

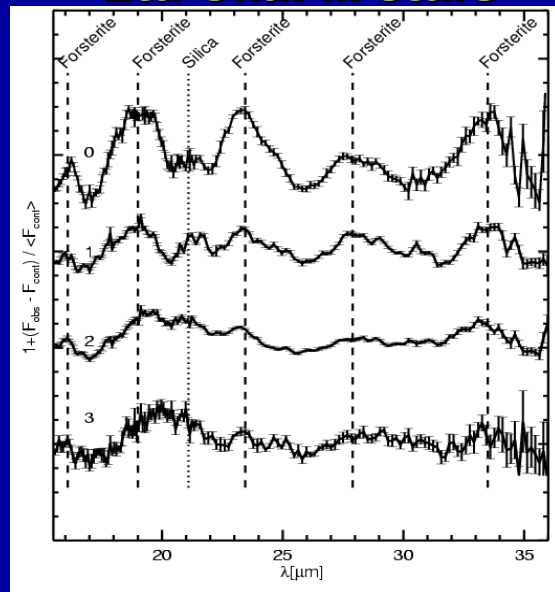
# Long-wavelength Observations of Disks

## FEPS:TTS



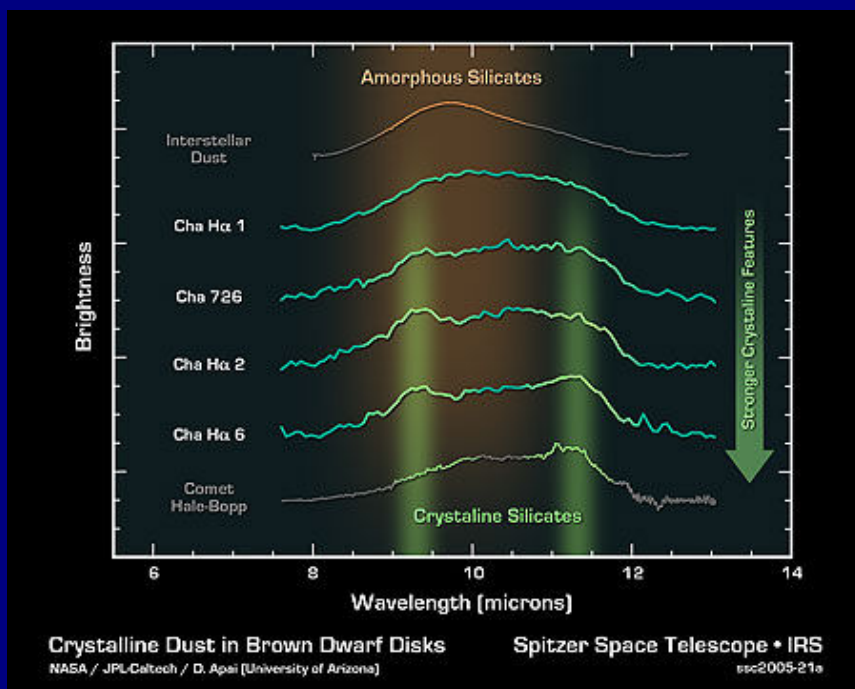
D=140 pc  
Age=5-10 Myr

## Eta Cha: M stars



D=100 pc  
Age=8 Myr

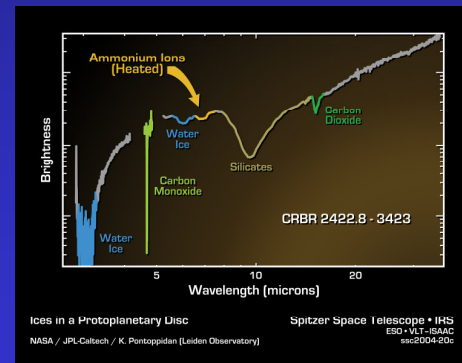
# Crystalline Dust in Brown Dwarf Disks



Apai et al. (Science, 2005)

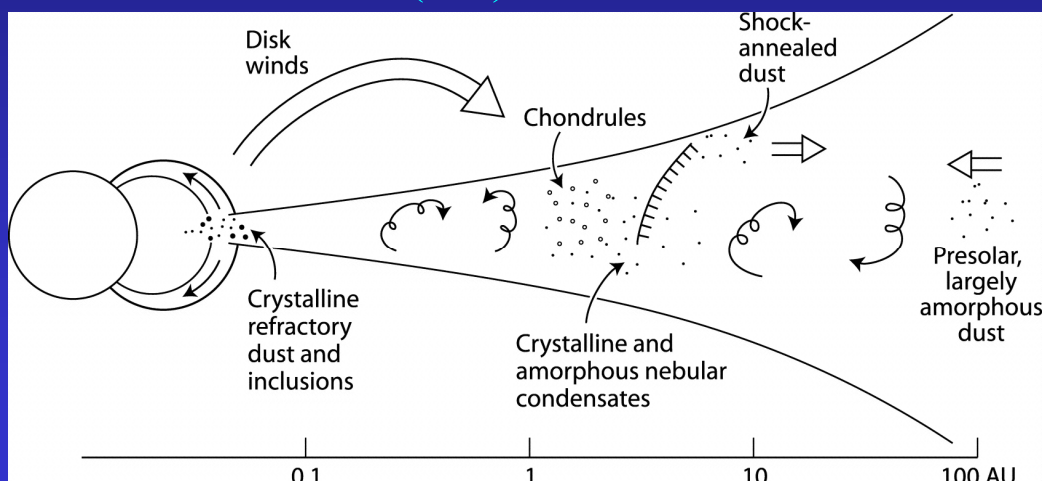
# Chemical composition

- PAHs in some of the systems
- Amorphous silicates present
- Mg-rich crystalline silicates exist (radial variation in structure)
- Silica exists
- No (strong) evidence for FeS
- No evidence for „organics“
- Evidence for simple molecular ices



# Origin of Crystalline Silicates in Protoplanetary Disks

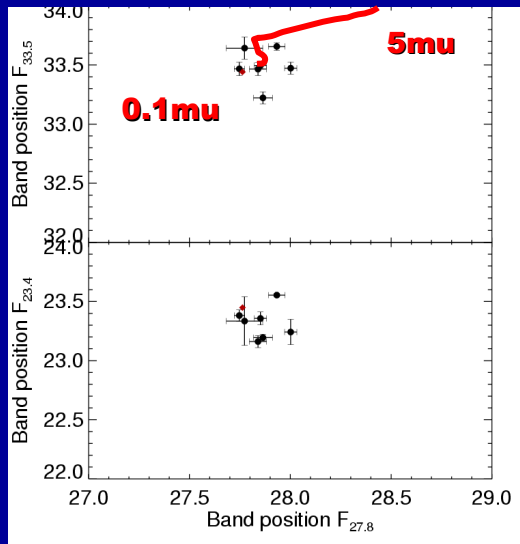
- Annealing/transport of amorphous silicates in/from inner disk or/and shock heating in the outer disk (annealing and/or condensation from the gas phase)
- Pre-solar stardust (???)



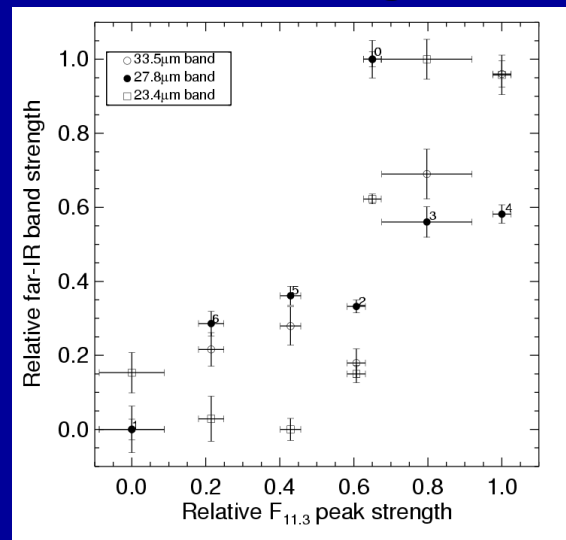
Scott & Krot (2005)

# The nature of the crystalline silicates

## Band position



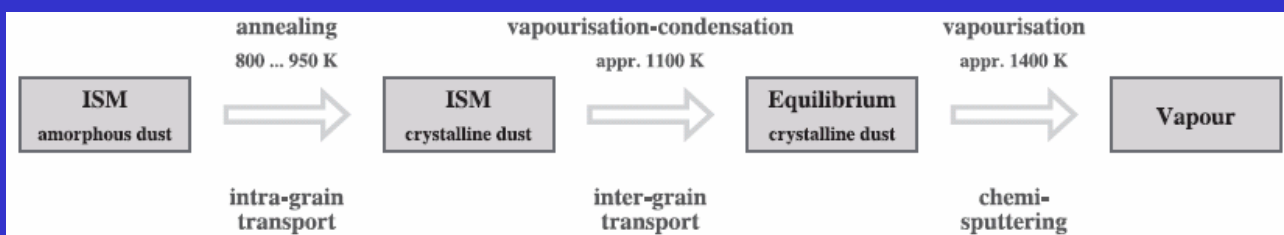
## Band strength



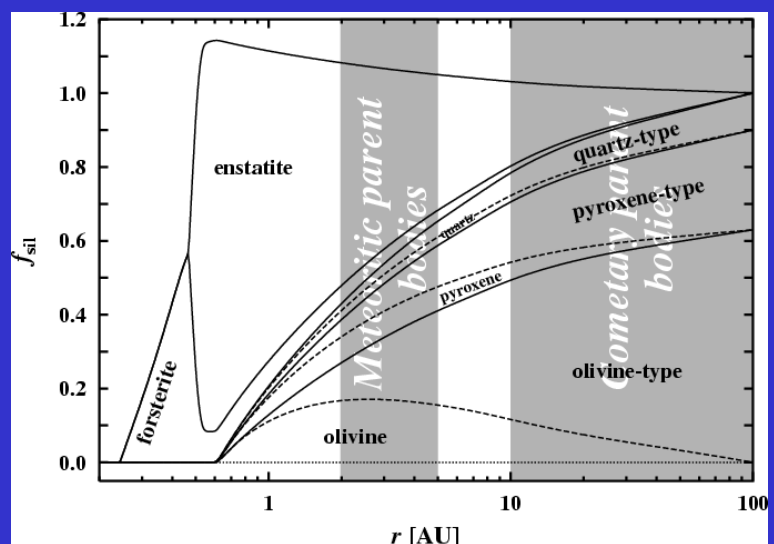
**Inner (1AU) and outer (~5-10AU) disk connected:  
Entire planet formation zone becomes crystalline**

**Crystalline silicates do not grow, but may get  
incorporated into larger aggregates (left)!**

# Mineralogy of Protoplanetary Disks

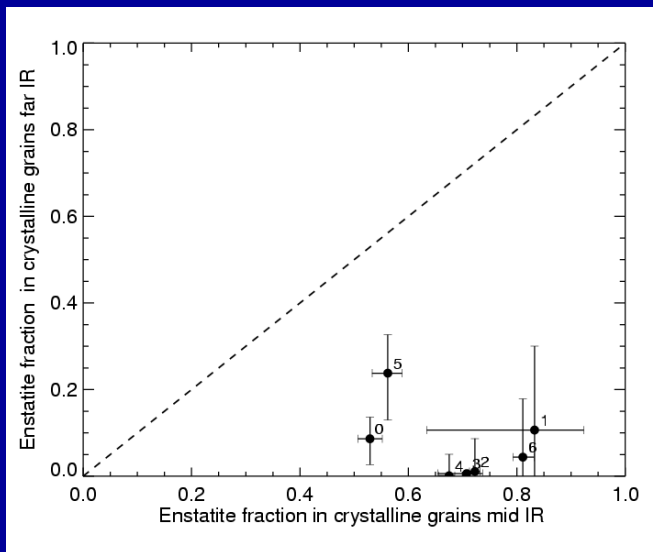


- ⇒ Different dust material in the inner and outer disk
- ⇒ Processing of (am.) quartz, olivine and pyroxene
- ⇒ Formation of crystalline Mg-rich end members and iron
- ⇒ Forsterite – dominant material in the inner disk



Gail (2004)

# Origin of crystalline silicates: Annealing or condensation/shocks?

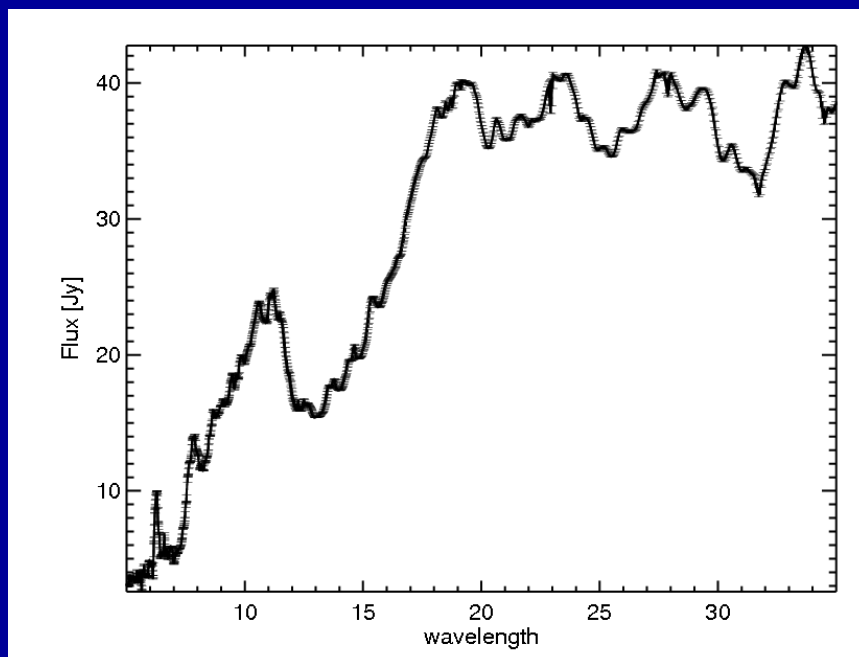


**TTS systems**

***Enstatite dominates in the inner disk (10  $\mu\text{m}$ )  
Forsterite in the outer disk (20-30  $\mu\text{m}$ ) !!***

**Bouwman, Henning et al. and FEPS team (2006)**

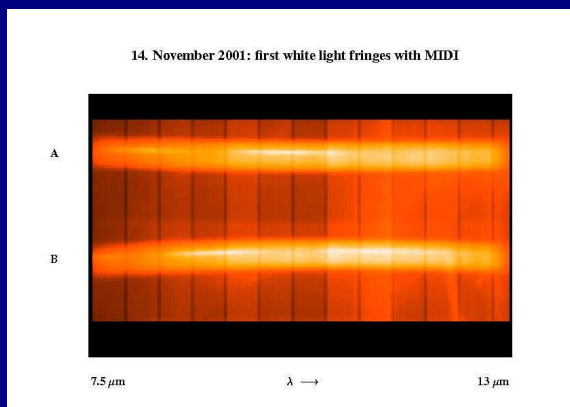
## **Enstatite in outer disk: The HAEBE star HD179218 exceptional case**



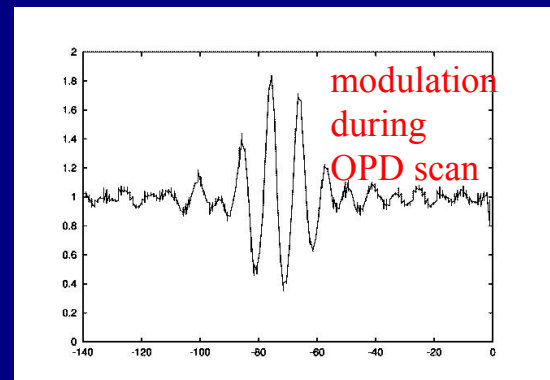
# Spectroscopy plus Interferometry - A New Frontier in Disk Studies -



Full detector



One wavelength bin

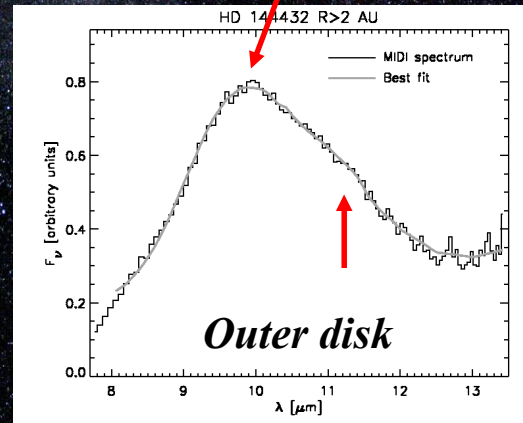
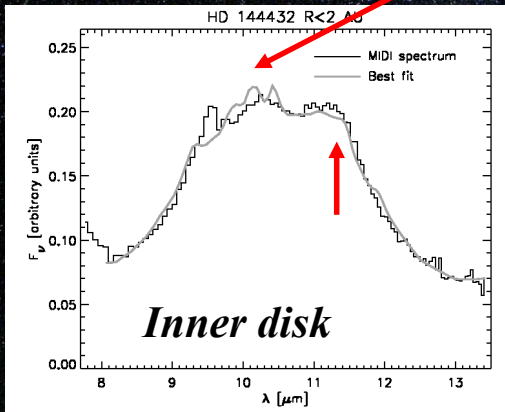
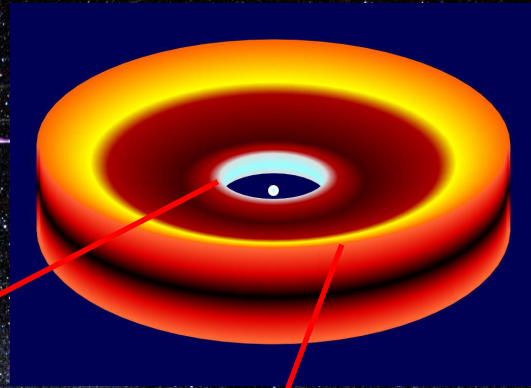
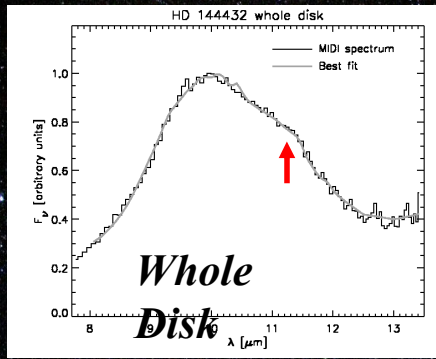


## The two aspects of dispersed interferometric measurements

Visibility  $V = \text{correlated flux} / \text{total flux} = I(x,y) \text{ at } (u,v) = \vec{B}_{\text{eff}} / \lambda$   
 van Cittert – Zernike – Theorem  
 → a size indicator

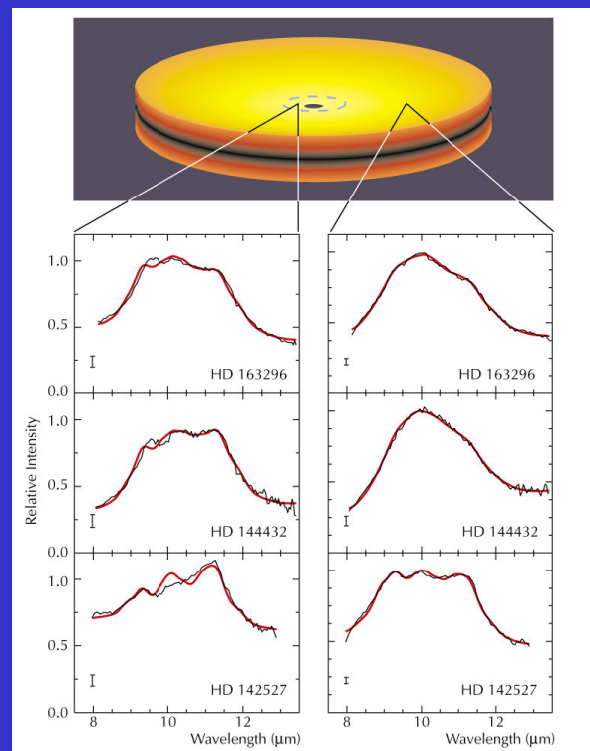
Correlated flux =  $V \cdot \text{total flux} = I_v$  of interferometrically selected region of characteristic size  $\lambda / B_{\text{eff}}$   
 → probes physical conditions

# Spatially resolved H<sub>Ae</sub> spectroscopy (MIDI)



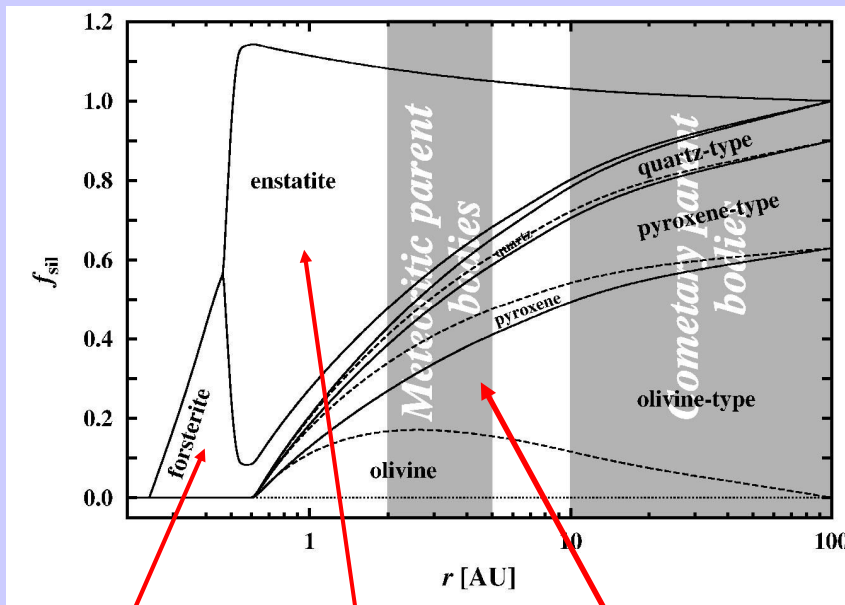
## Spatial distribution of the dust

- Crystalline grains concentrated in central disk regions
- Outer disks can be “pristine” while inner disks are “evolved”
- In disks with low crystallinity, crystals seem restricted to innermost disk region
- In disks with high crystallinity, crystals are present also further out.
- HD 142527: inner disk mostly forsterite, further out more enstatite





# Radial Distribution of Silicates

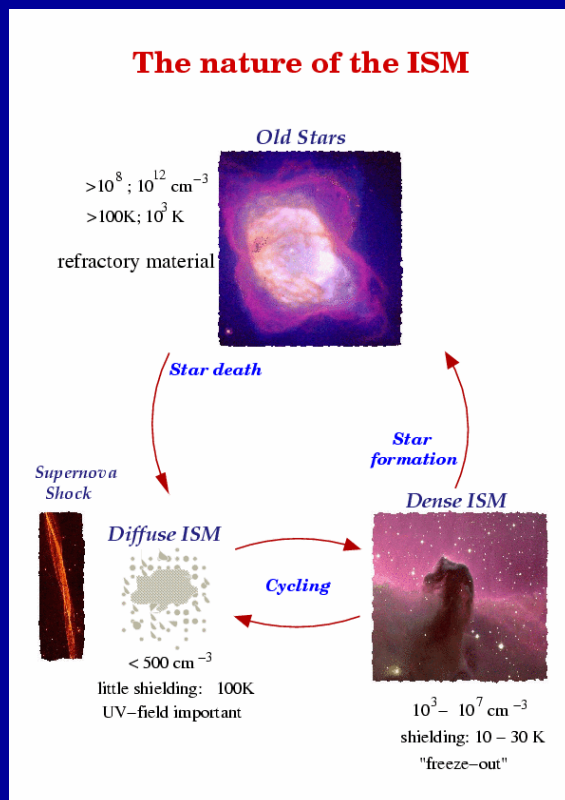


Gail  
(2004)

Forsterite Enstatite

Fe-rich amorphous  
olivines, pyroxenes

## The Lifecycle of Dust: Crystalline vs. Amorphous Dust

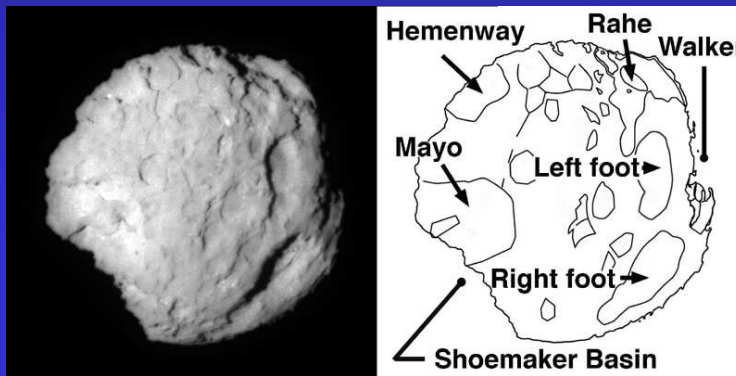


crystallinity  $x$

Evolved (AGB, PN, RSG)	11-18 %
Evolved (SN)	?
diffuse ISM	<0.4 %
Star-forming regions	Small
Herbig Ae/Be, T Tau stars	5-8 %
Debris disks	?
Solar system	Very high

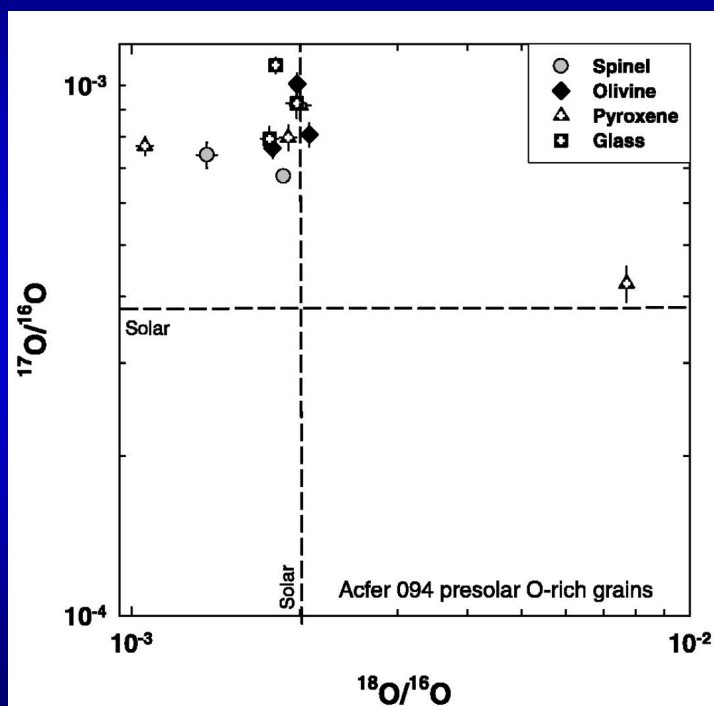
# In-situ studies of primitive material in the solar system

- Detailed chemical, mineralogical, and isotopic analysis
- Properties of the „initial“ grain population
- Comparative studies with protoplanetary disks



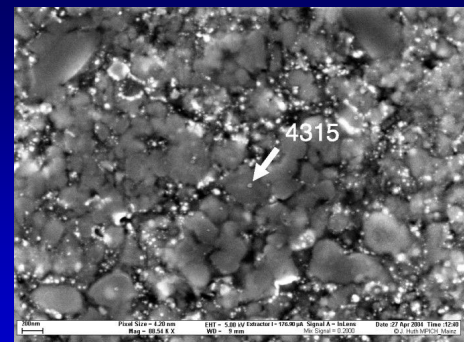
Stardust probe  
Closest encounter  
July 2, 2004

## Silicates from Space



AGB or RGB origin

SEM Image



Scale bar: 200 nm

- 3 Olivine grains
- 4 Pyroxene grains
- 3 Glass-like grains

Hoppe et al. 2005

(see also Messenger et al. 2003)

# Glass with embedded metals and metal sulphides

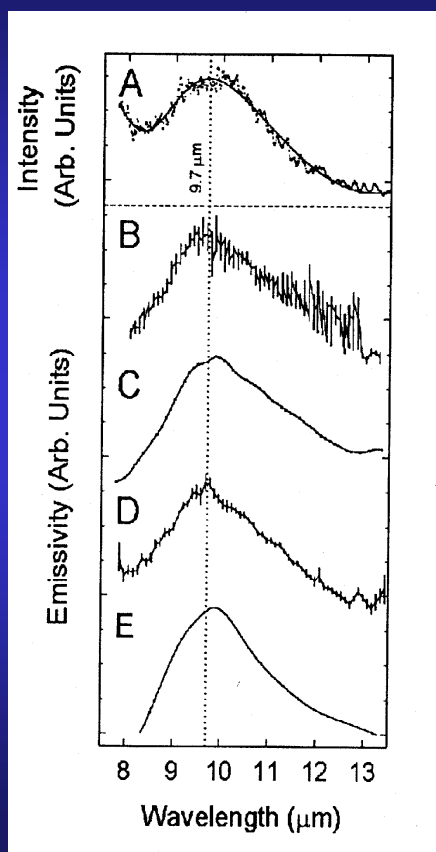
## GEMS



Bradley (1994)

- Abundant component of anhydrous interplanetary dust particles (IDPs)
- Silicate glasses with inclusions of iron-nickel metal and iron-rich sulfides
- Extended ion irradiation history

## Comparison of the 10 $\mu\text{m}$ Si-O stretch band



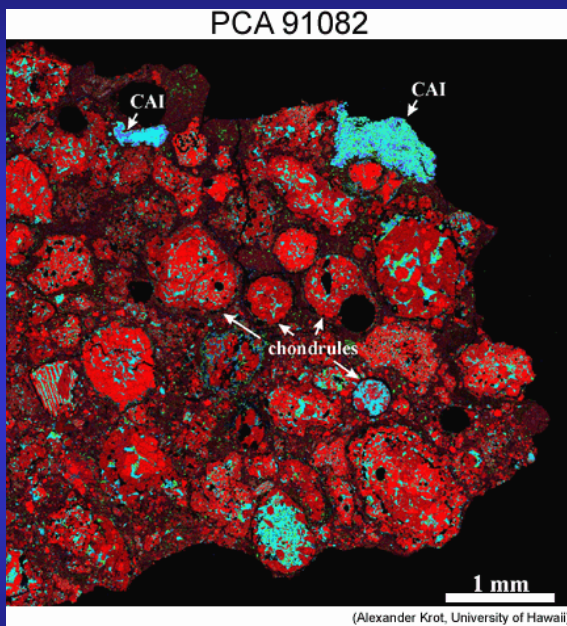
Spectral ambiguity ....

- A** GEMS in IDP L2011\*B6
- B** Elias 16
- C** Trapezium
- D** DI Cep (T Tauri star)
- E**  $\mu$  Cep ( M supergiant)

GEMS:  $(\text{Mg}+\text{Fe})/\text{Si}\sim 0.7$   
(Keller & Messenger 2004)

Bradley et al. (1999)

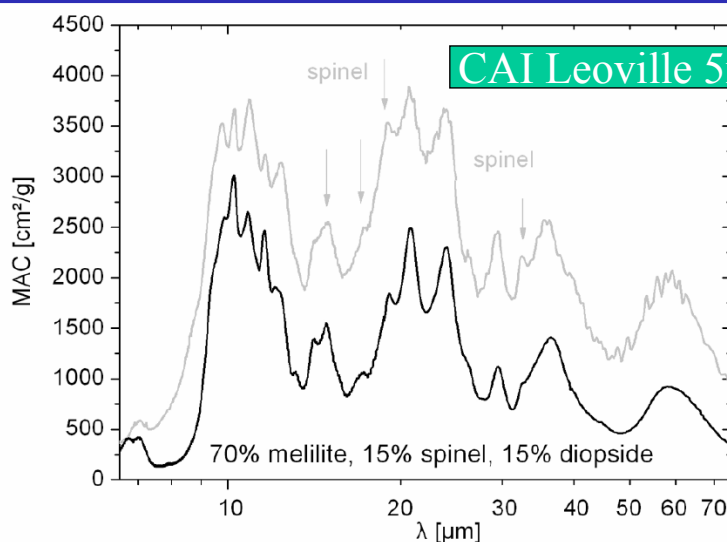
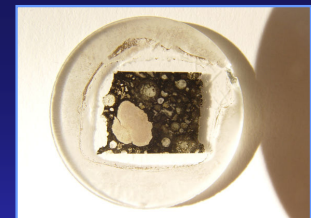
# Calcium-Aluminium Rich Inclusions (CAIs)



- Oldest known objects in the solar system; age 4.57 billion years (Amelin et al. 2002)
- Consist of high-temperature minerals

X-ray elemental map  
Mg – red, Ca – green, Al - blue

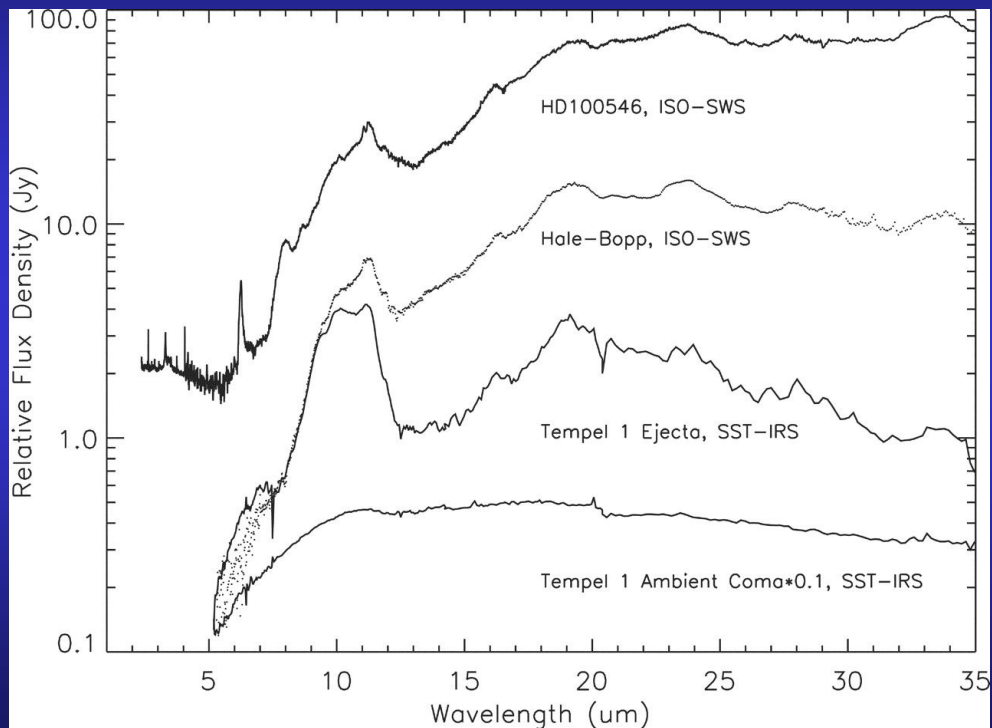
## Spectroscopy of CAIs



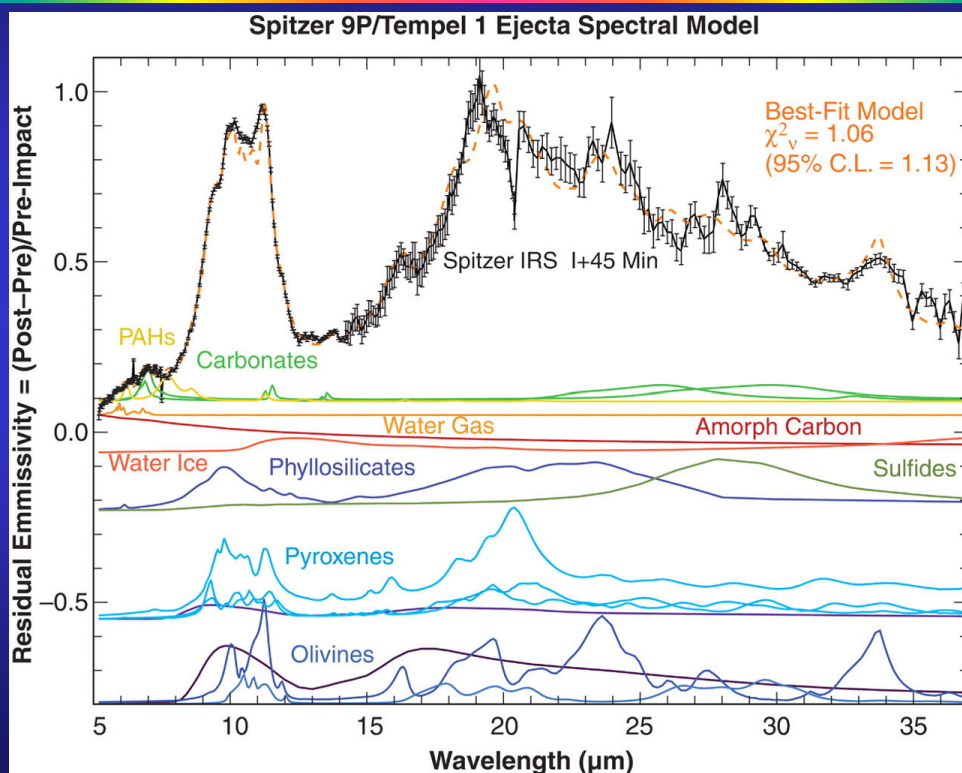
- Diopside  
 $\text{CaMgSi}_2\text{O}_6$
- Spinel  
 $\text{MgAl}_2\text{O}_4$
- Melilite –  
Solid solution of  
gehlenite  
( $\text{Ca}_2\text{Al}_2\text{SiO}_7$ )  
åkermanite  
( $\text{Ca}_2\text{MgSi}_2\text{O}_7$ )

Posch et al. (2006, submitted)

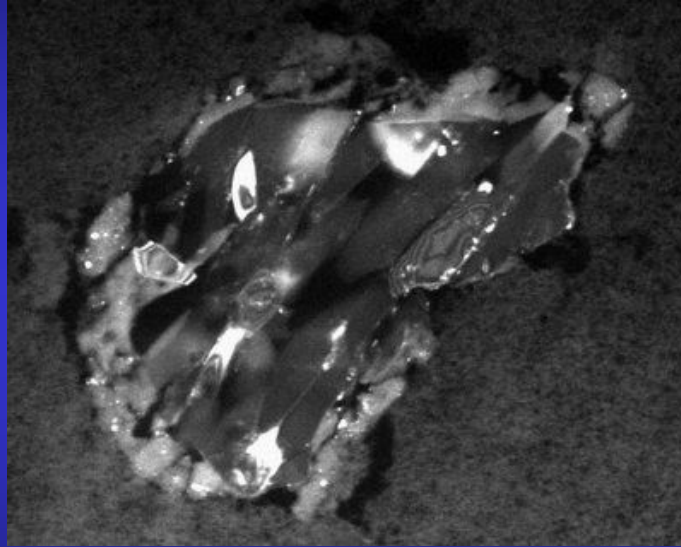
# Spitzer Results – Deep Impact Ejecta of Comet 9P/Tempel 1



# Spitzer Results – Deep Impact Ejecta of Comet 9P/Tempel 1



## First Stardust Results – Mg-rich olivine crystals from comet Wild 2



Scale  
2 microns

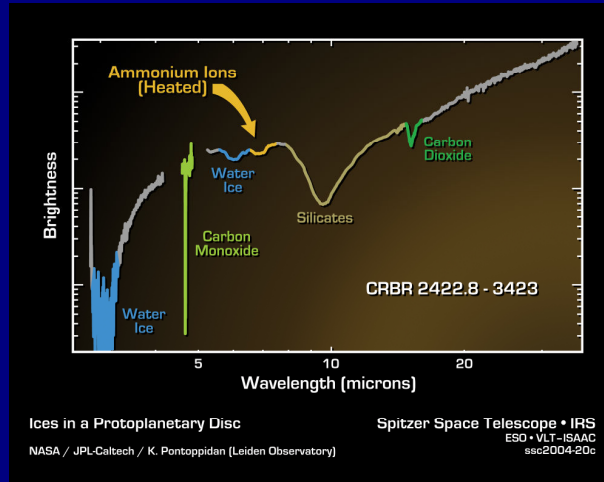
Other high-temperature crystals rich in Ca, Al, Ti

## Summary

- Amorphous silicates present
- Crystalline Mg-rich olivine and pyroxene detected
- No evidence for iron sulphides
- Spatially resolved data are becoming available
- In-situ study of primitive material in solar system

# Open questions

- How are the crystalline silicates produced?
- Do we see other high-temperature solids?
- Where is the iron?
- What is the structure of grain mantles?



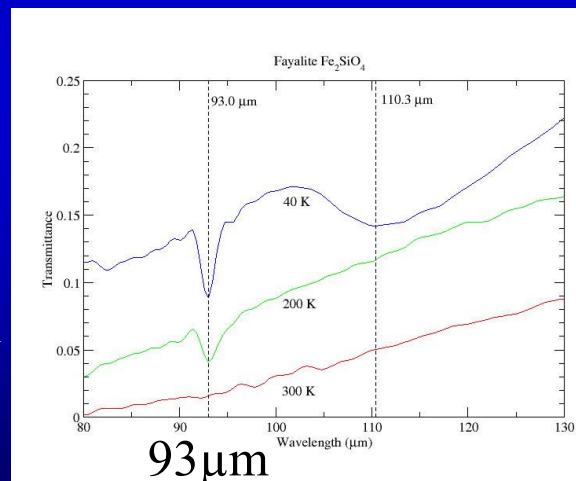
Stay tuned for Spitzer and Stardust results ...

## Herschel - The future ?

FIR: Lattice vibrations of heavy ions or ion groups with low bond energies (example KBr: transverse optical mode at 86  $\mu\text{m}$ )

PACS: 57-210  $\mu\text{m}$

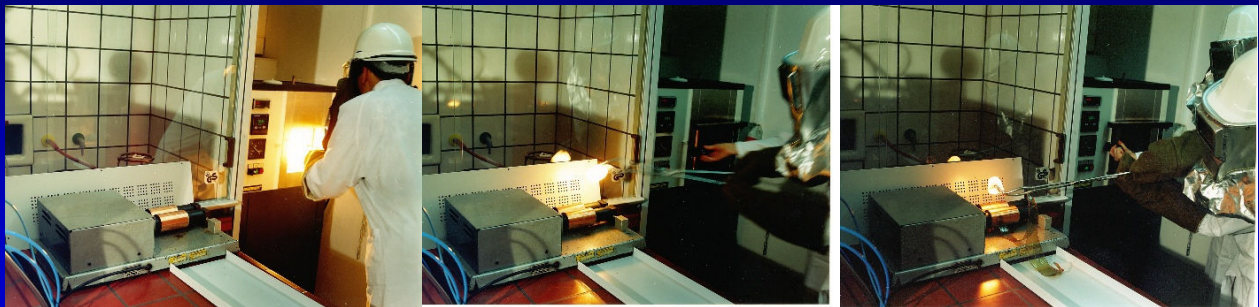
- Forsterite 69  $\mu\text{m}$  band
- Fayalite 93-94  $\mu\text{m}$  and 110  $\mu\text{m}$  band
- Hydrous silicates 100-110  $\mu\text{m}$  (e.g. montmorillonite)



Absorption, scattering, and emission by interstellar material produces enough puzzles, even of identification, to keep the proverbial seven spectroscopists with seven brooms busy for at least seven years.

Trimble & Aschwaden (1998)

## Collaborators and Reviews



**F. Huisken, C. Jäger, H. Mutschke (Lab AIU Jena/MPIA Heidelberg)**

**R. van Boekel, J. Bouwman (ISO/Spitzer/MIDI, MPIA Heidelberg)**

### Reviews:

Henning, Th. (ed.): *Astromineralogy*. Springer 2003.

Henning, Th., Mutschke, H.: Optical Properties of Cosmic Dust Analogs. In: *ASP Conf Series*. 2000. p.253

Colangeli, L., Henning, Th. et al.: The Role of Laboratory Experiments in the Characterization of Silicon-based Materials. *AA Rev.* 11, 97, 2003.

Henning, Th., Mutschke, H., Jäger, C.: Silicates – Space and Laboratory. *Astrochemistry. Recent Successes and Current Challenges*. IAU Symp. 231. 2006.